

Polydopamine photochemical behavior under UV irradiation

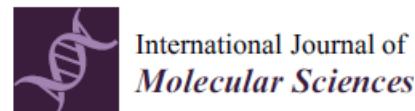
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- 3 Photochemistry of polydopamine polymer (trimer) chains
- 4 Conclusions

Dopamine photochemistry



Int. J. Mol. Sci. 2022, 23, 5483



Article

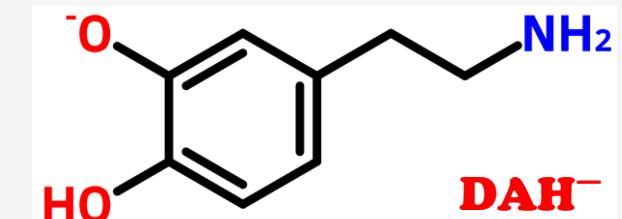
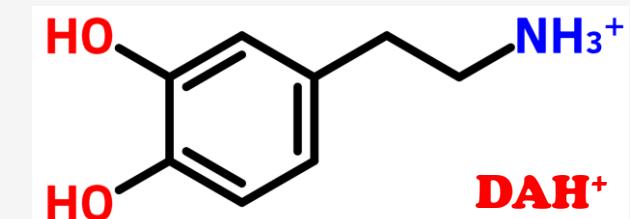
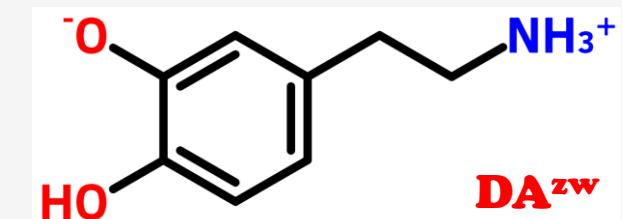
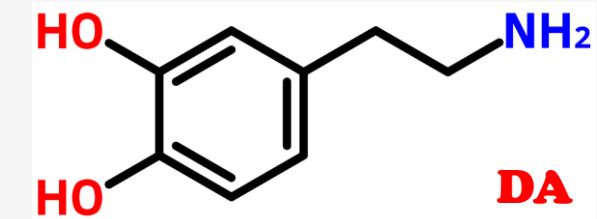
Dopamine Photochemical Behaviour under UV Irradiation

Alexandra Falamaş ¹, Anca Petran ¹, Alexandru-Milentie Hada ^{2,3} and Attila Bende ^{1,*}

Experimental and theoretical investigation:

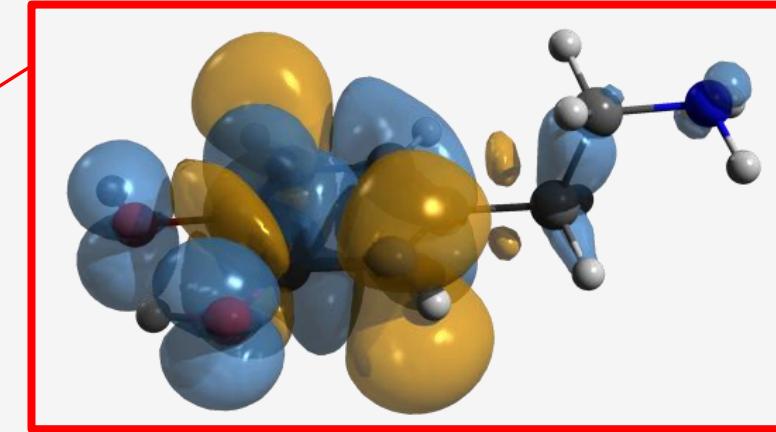
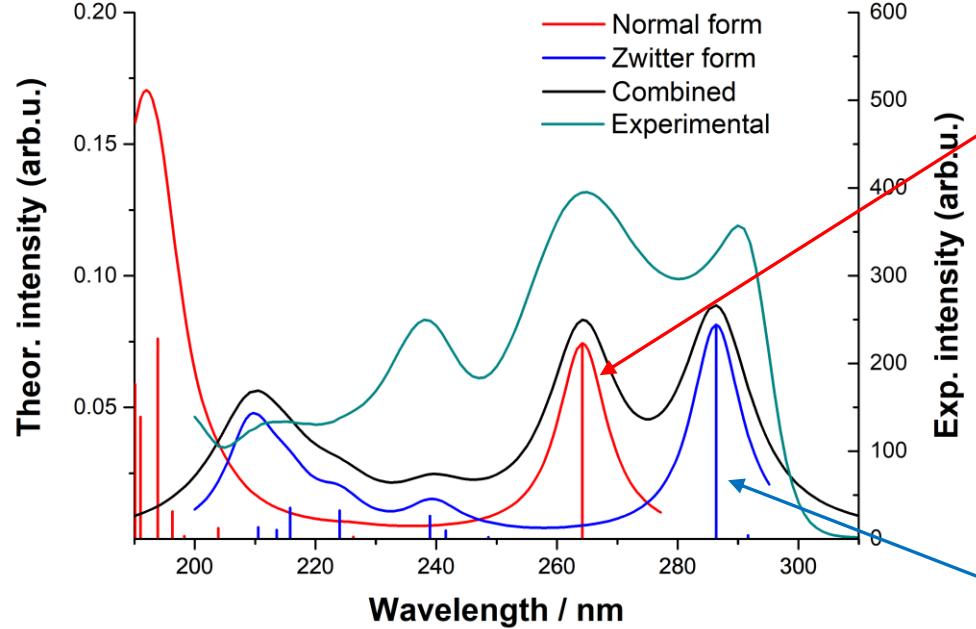
DA, DA^{zw}, DAH⁺, and DAH⁻

- UV-absorption and Fluorescence emission spectra;
- Fluorescence Lifetimes and Quantum Yield;



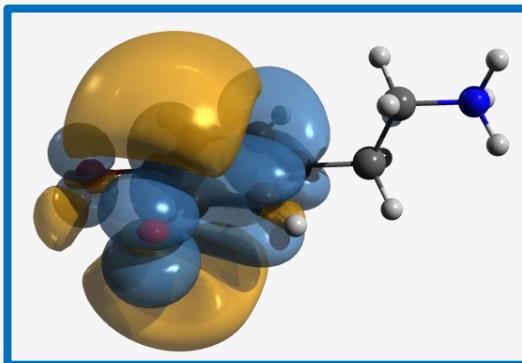
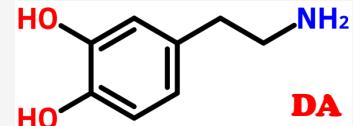
Dopamine photochemistry

UV-absorption spectra:



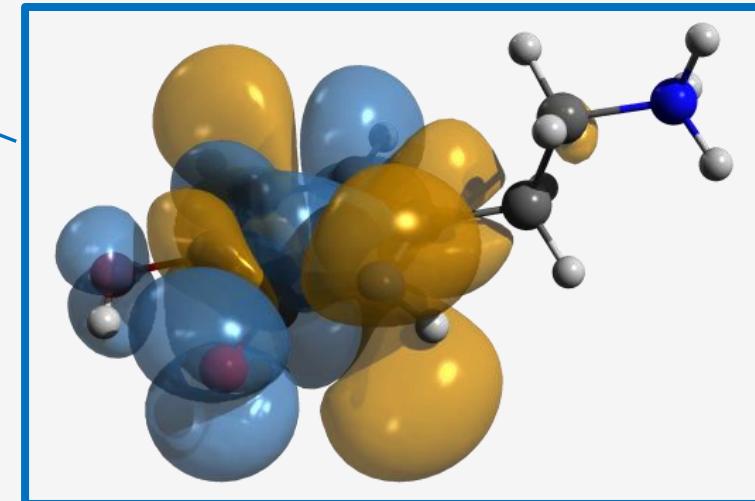
$S_0 \rightarrow S_1$

$n \rightarrow \pi$



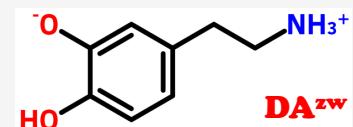
$S_0 \rightarrow S_1$

Rydberg state !!!



$S_0 \rightarrow S_2$

$n \rightarrow \pi$



Dopamine photochemistry

Electronic excited states:

Table 1. Vertical electronic transitions (in nm) and their oscillator strengths in the spectral range of 200–300 nm found for the DA, DA^{zw}, DAH⁺, and DAH⁻ equilibrium geometries obtained at the DLPNO-STEOM-CCSD/ma-def2-TZVPP level of theory.

State	Geometries								
	DA		DA ^{zw}		DAH ⁺		DAH ⁻		
	ω	f		ω	f		ω	f	
S ₁	266	0.0275		292	0.0014		270	0.0379	
S ₂	226	0.0024		286	0.0803		222	0.0011	
S ₃	208	0.0637		249	0.0008		209	0.0378	
S ₄				242	0.0033			250	0.0164
S ₅				239	0.0088			230	0.0947
S ₆				224	0.0108			226	0.1094
S ₇				216	0.0119			219	0.0050
S ₈				214	0.0034			216	0.0018
S ₉				211	0.0045			216	0.0107
S ₁₀				209	0.0351			211	0.0002

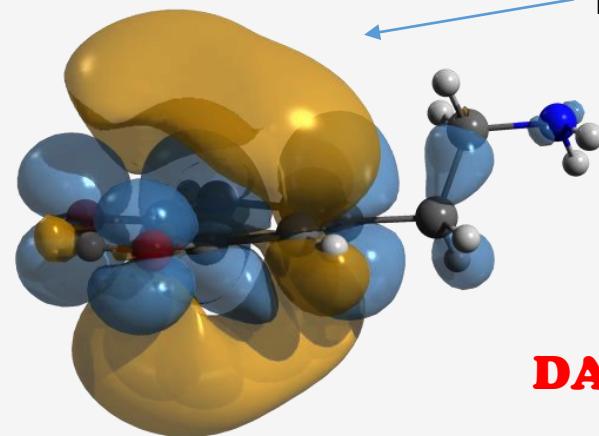
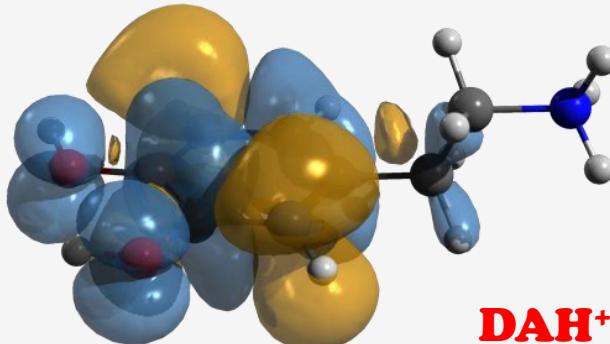
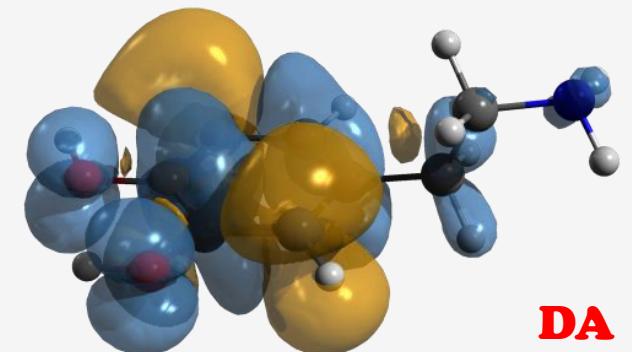
Small oscillator strength (f) for S₁ state in case of DA^{zw} and DAH⁻

Dopamine photochemistry

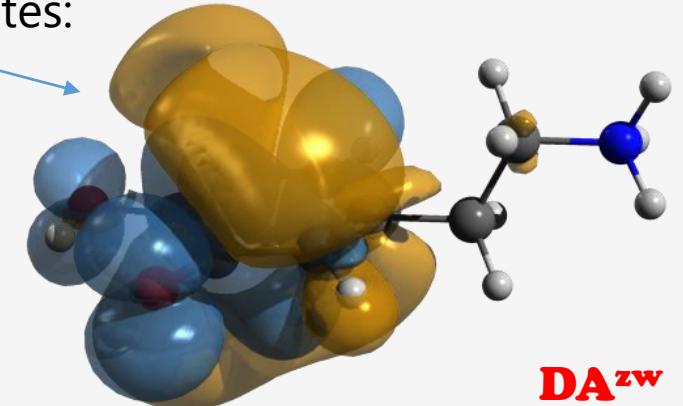
Absorption - Emission:

	DA		DA ^{zw}		DAH ⁺		DAH ⁻	
	λ^{abs}	λ^{em}	λ^{abs}	λ^{em}	λ^{abs}	λ^{em}	λ^{abs}	λ^{em}
ω B97X-D3	240.3	269.9	274.3	317.2	239.4	268.9	282.7	323.8
DLPNO-STEOM-CCSD	265.5	301.4	291.7 (286.4)	334.3	269.7	299.6	302.2 (286.8)	340.1
Exp.	280	317	280	317	280	317	280	317

$S_1 \rightarrow S_0$



Rydberg states:



Dopamine photochemistry

Excited state lifetime (experimental):

Table 2. The parameter fits (fluorescence lifetime components, errors, relative amplitude, and amplitude average lifetime) calculated using the reconvolution method for the concentrated, respectively diluted, solutions at both pH values.

pH	Solution	Reconv. Method	τ_1 (ns)	Error	Ampl. (%)	τ_2 (ns)	Error	Ampl. (%)	$\tau_{avr.}$ (ns)	Error
5.5	Concentrated	1-exp	0.89	0.01	100	-	-	-	-	-
		2-exp	0.91	0.02	99.58	4.86	0.38	0.42	0.99	0.03
	Diluted	1-exp	0.93	0.07	100	-	-	-	-	-
	Concentrated	2-exp	0.89	0.01	95.83	2.98	0.20	4.17	0.97	0.05
8.0		2-exp	0.71	0.10	88.97	3.23	0.24	11.03	1.56	0.15
Concentrated	2-exp	0.69	0.02	92.61	2.41	0.20	7.40	0.82	0.01	
	2-exp	0.64	0.10	88.20	2.96	0.18	11.80	0.91	0.11	

Dopamine photochemistry

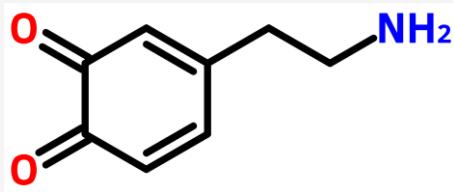
Excited state
lifetime
(theoretical):

	DA	DA^{zw}	DAH⁺	DAH⁻
FC ^a	4.39×10^{-9} (100%)	8.47×10^{-7} (100%)	4.70×10^{-9} (100%)	1.36×10^{-6} (100%)
FC+HT ^b	3.79×10^{-9} (86%, 14%)	2.75×10^{-9} (0%, 100%)	4.51×10^{-9} (96%, 4%)	1.14×10^{-8} (1%, 99%)
FC+(HT+D ^c)	3.16×10^{-9} (81%, 19%)	1.02×10^{-9} (0%, 100%)	2.95×10^{-9} (78%, 22%)	7.10×10^{-9} (0%, 100%)

^a Franck–Condon approximation. ^b Herzberg–Teller approximation. ^c Duschinsky rotation effects.

pH	Solution	QY (%)	
5.5	Concentrated	3.4	Quantum Yield
	Diluted	3.4	
8.0	Concentrated	2.0	
	Diluted	2.4	

Photochemistry of polydopamine constituents



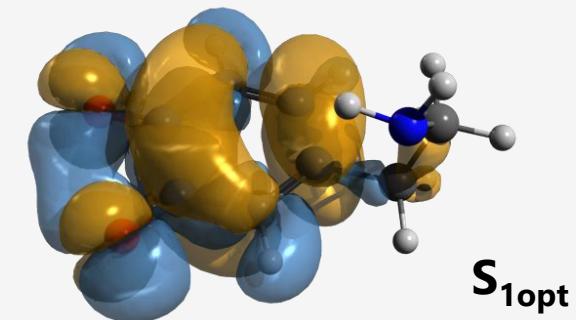
Dopamine o-quinone

Electronic excited states:

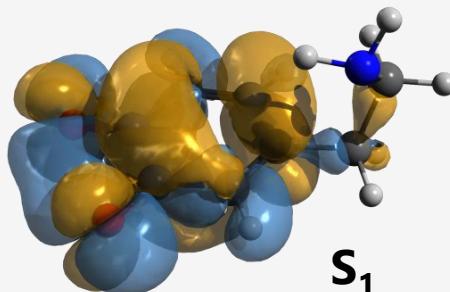
	ωB97X-D3	DLPNO-STEOM-CCSD
S₁	512.3	581.3
S₂	366.2	375.0
S₃	322.9	349.2
S₄	277.4	248.8
S₅	213.5	213.8

Fluorescence emission:

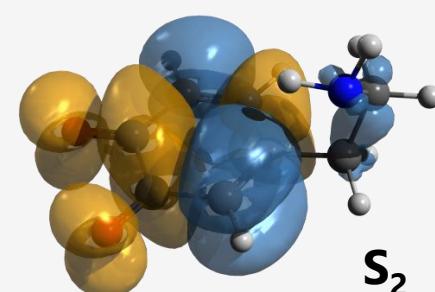
	ωB97X-D3	f
S₁	607.6	0.0003
S₂	403.1	0.1112



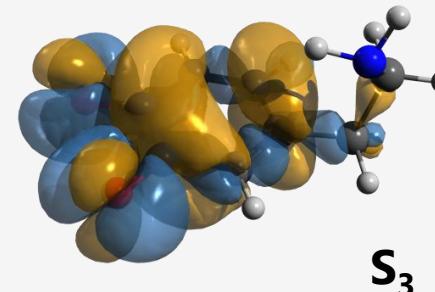
S₁opt



S₁



S₂



S₃

Fluorescence lifetime:

$$\tau = 3.55 \mu\text{s}$$

Photochemistry of polydopamine polymer (trimer) chains

Excited state geometry optimization with DFT – Problems with the exchange correlation functionals

They were optimized for ground electronic states !!!

Lack of electron – hole interaction (charge transfer effects)

Reference method: EOM-CCSD/def2-TZVP(-f)

	$\Delta(\text{C-C arom})$	$\Delta(\text{C-O})$	$\Delta(\text{C-C})$	$\Delta(\text{C-C alif})$	$\Delta(\text{C-N})$	Total
ωB97X	0.00617	0.00525	0.00370	0.00090	0.00780	0.0048
$\omega\text{B97X-D3}$	0.00628	0.00605	0.00440	0.00140	0.00830	0.0053
$\omega\text{B97X-D3BJ}$	0.00470	0.00090	0.00110	0.00150	0.00230	0.0021
$\omega\text{B97X-V}$	0.00365	0.00040	0.00240	0.00460	0.00150	0.0025

Values are given in Å

$\omega\text{B97X-D3BJ}$ can reproduce quite well the EOM-CCSD results !!!

Photochemistry of polydopamine polymer (trimer) chains

How accurate can DFT functionals reproduce the STEOM-DLPNO-CCSD results of excited state energies?

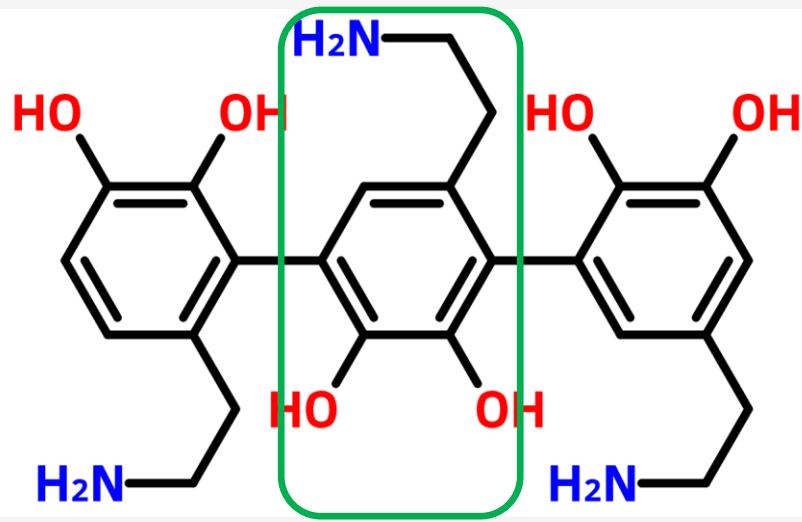
	ΔS_1 (nm)	ΔS_2 (nm)	ΔS_3 (nm)	ΔS_4 (nm)	Average (nm)
B3LYP-D3BJ	11.0	-27.0	-23.0	-23.0	21
CAM-B3LYP-D3	21.0	-7.0	-1.0	-6.0	8.75
revPBE0-D3	16.0	-17.0	-14.0	-16.0	15.75
MN15	19.0	14.0	-2.0	9.0	11
TPSS-D3	-5.0	-38.0	-42.0	-50.0	33.75
TPSSh_05-D3	3.0	-31.0	-33.0	-39.0	26.5
TPPSH_10-D3	8.0	-25.0	-26.0	-29.0	22
TPPSH_20-D3	16.0	-14.0	-13.0	-16.0	14.75
TPPSH_25-D3	20.0	-9.0	-7.0	-12.0	12
ωB97X-D3	24.0	8.0	2.0	6.0	10
ωB97X-D3BJ	26.0	7.0	3.0	6.0	10.5
ωB97X-D4	26.0	7.0	3.0	6.0	10.5
mPW2PLYP	14.0	-21.0	-9.0	-15.0	14.75
PBE-QIDH	20.0	4.0	4.0	2.0	7.5
wB2PLYP	23.0	1.0	4.0	4.0	8
RSX-QIDH	28.0	14.0	7.0	13.0	15.5
wB97X-2	-20.0	-17.0	-4.0	-6.0	11.75
SCS-PBE-QIDH	7.7	3.1	0.5	1.3	3.15
SOS-PBE-QIDH	4.5	3.6	-0.5	1.8	2.6

Reference method: DLPNO-STEOM-CCSD/ma-def2-TZVPP

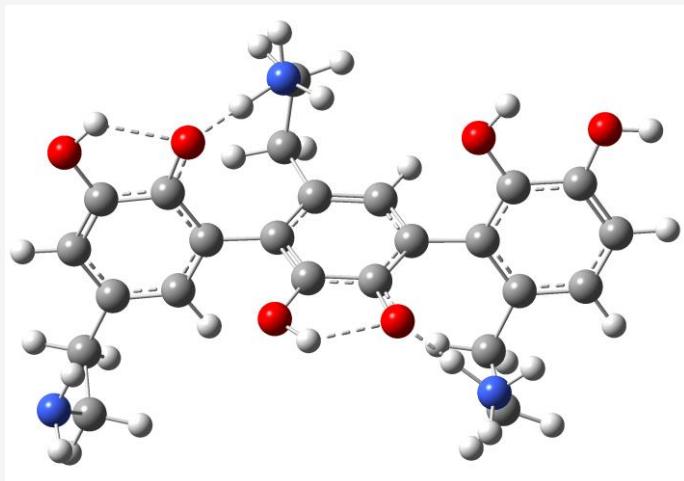
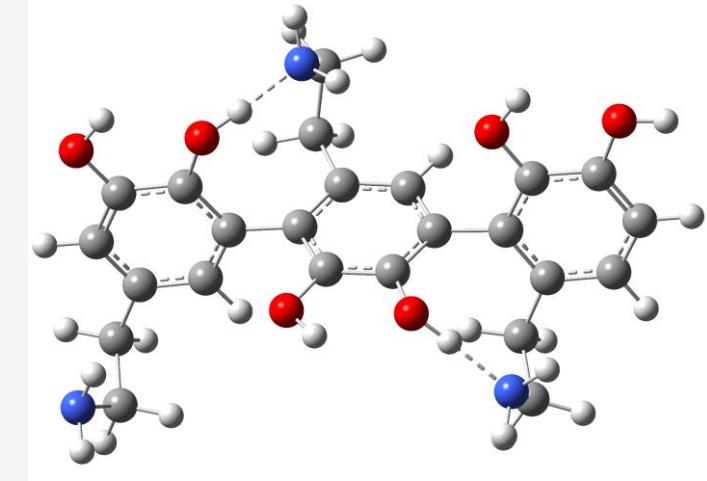
Geometry optimization:
with **ω B97X-D3BJ**

Excited state energies:
with **SOS-PBE-QIDH**

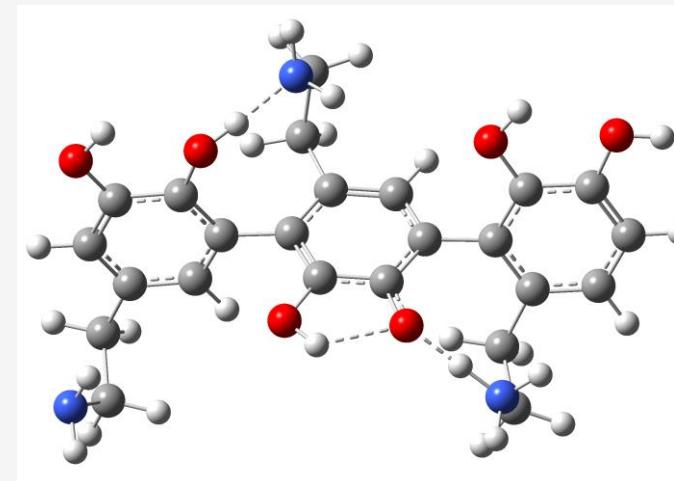
Photochemistry of polydopamine polymer (trimer) chains



DA-DA-DA
DA-DA^{zw}-DA
DA-DA^{zw}-DA^{zw}



2.64 kcal/mol



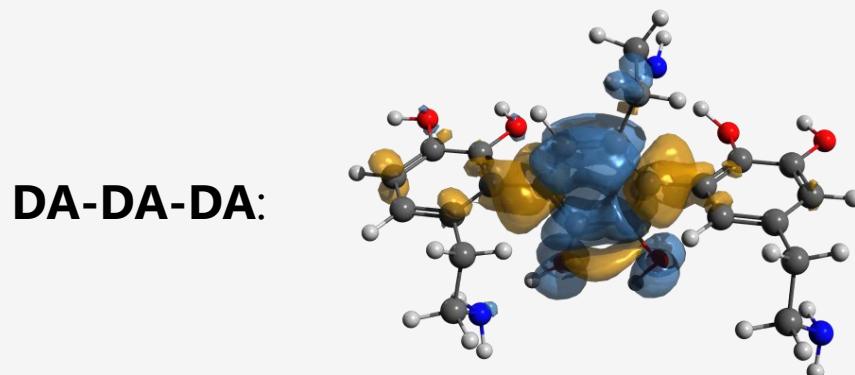
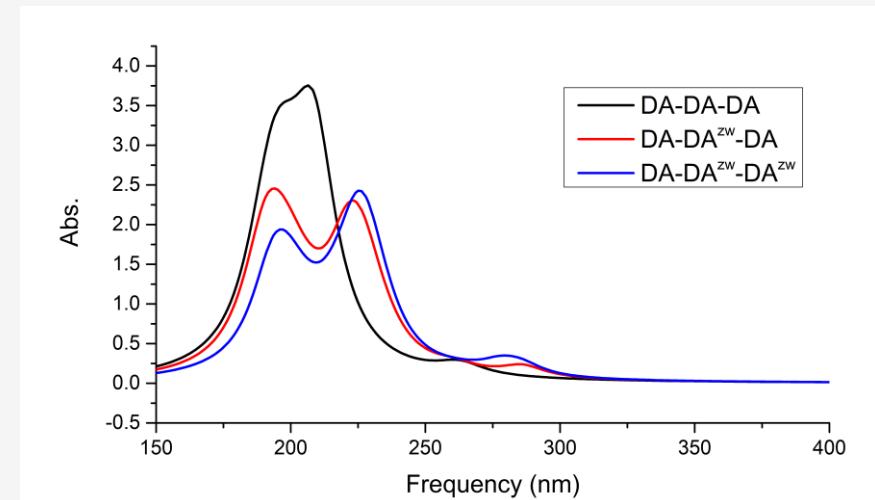
2.37 kcal/mol



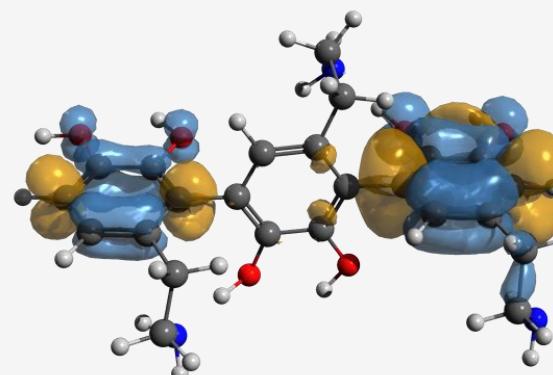
Photochemistry of polydopamine polymer (trimer) chains

Electronic excited states (SOS-PBE-QIDH/ma-def2-TZVPP)

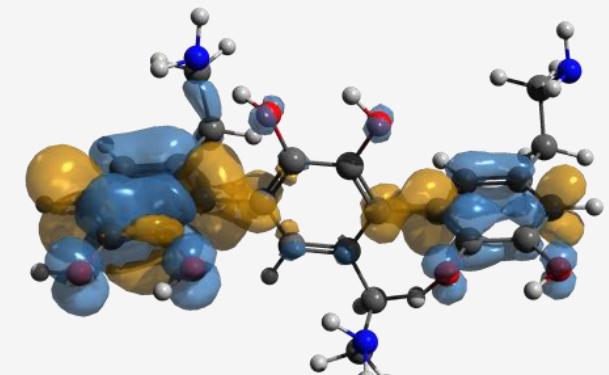
	DA (nm)	DA ^{zw} (nm)	DA-DA-DA (nm)	DA-DA ^{zw} -DA (nm)	DA-DA ^{zw} -DA ^{zw} (nm)
S ₁	263.5	284.2	265.1	286.1	285.5
S ₂	224.4	237.2	261.4	261.7	277.9
S ₃	211.5	204.5	261.3	260.6	261.7
S ₄	200.2	207.5	216.8	236.3	237.0
S ₅	197.6	194.0	213.5	230.6	230.6
S ₆	193.1	212.9	212.4	229.8	226.7



$S_0 \rightarrow S_1$



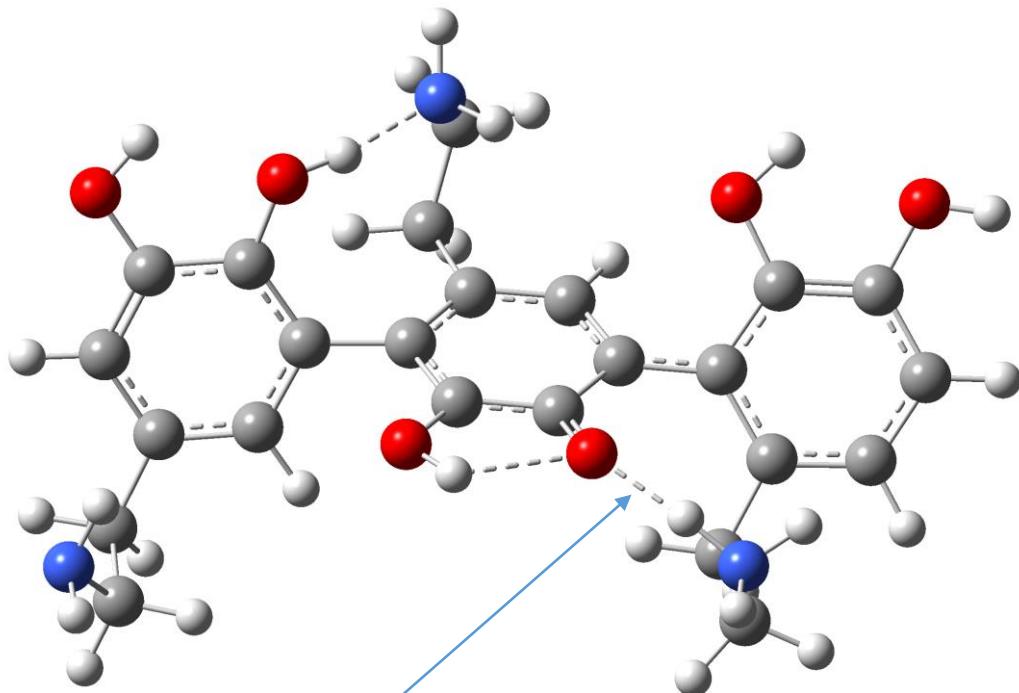
$S_0 \rightarrow S_2$



$S_0 \rightarrow S_3$

Photochemistry of polydopamine polymer (trimer) chains

DA-DA-DA's S_1 optimized geometry:



Internal proton transfer can easily occurs during the excited state relaxation !!!

Fluorescence emission wavelength:

DA:

$$\lambda^{\text{em}} = 301.4 \text{ nm}$$

DA-DA-DA:

$$\lambda^{\text{em}} = 419.3 \text{ (324.4)} \text{ nm}$$

DA-DA^{zw}-DA:

$$\lambda^{\text{em}} = 419.4 \text{ nm}$$

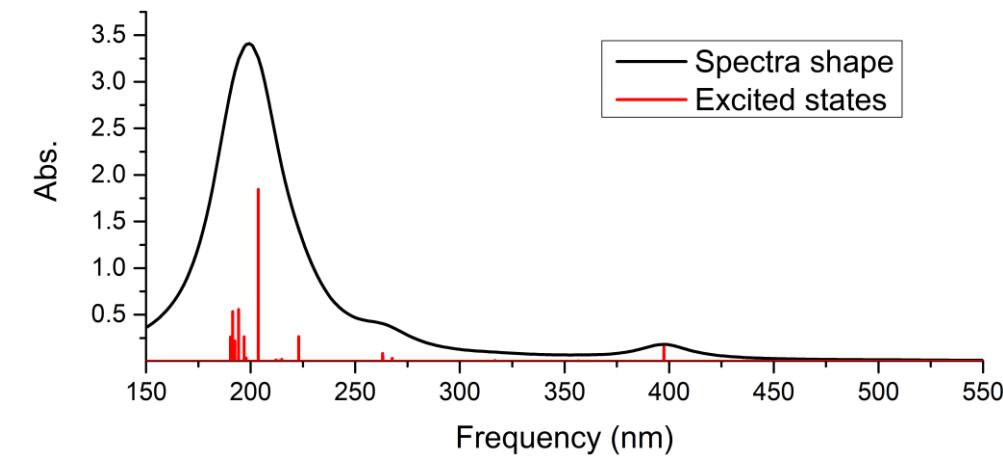
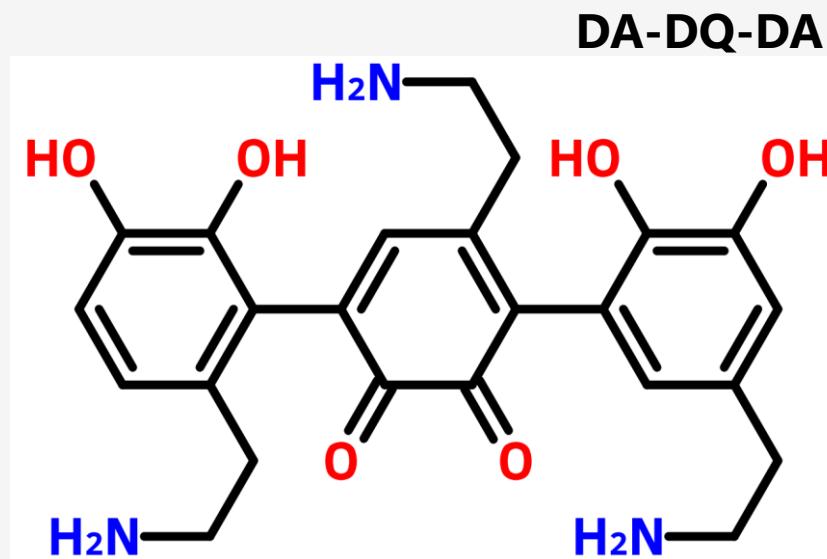
Fluorescence lifetime:

DA: $\tau = 3.16 \text{ ns}$

DA-DA-DA: $\tau = 7.10 \text{ ns}$

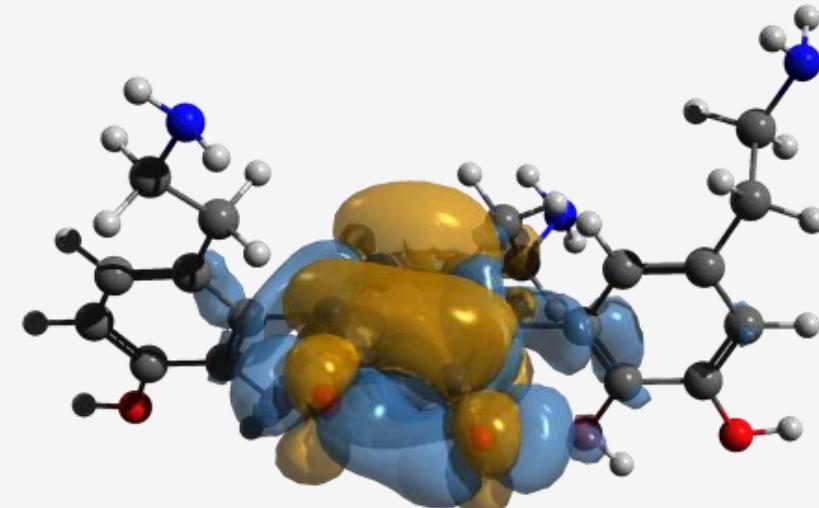
DA-DA^{zw}-DA: $\tau = 17.84 \text{ ns}$

Photochemistry of polydopamine polymer (trimer) chains



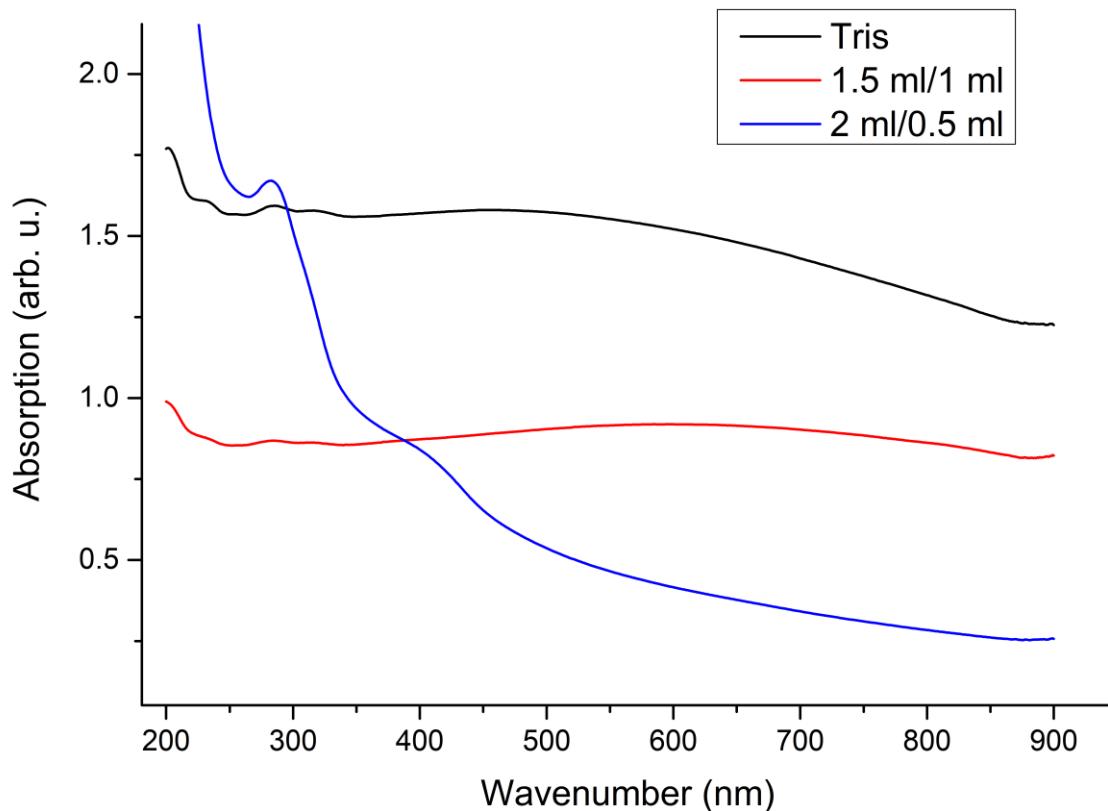
State	ω B97X-D3BJ	SOS-PBE-QIDH
S_1	481.1	506.5
S_2	379.9	397.5
S_3	334.6	439.4
S_4	329.1	423.2
S_5	313.6	356.6
S_6	295.2	331.6
S_7	283.8	330.3
S_8	270.7	316.7
S_9	239.8	263.2
S_{10}	238.9	263.1

$$\lambda^{\text{em}} = 618 \text{ nm}$$



Photochemistry of polydopamine polymer (trimer) chains

Experimental UV-Vis spectra:



Different structural forms can be seen for low and high concentrations

Large band shape due to several slightly different structures (for ex. the position of protonation or deprotonation as well as supramolecular aggregation)

Conclusions:

- Dopamine shows different photochemical behavior depending on its structural configuration (neutral, zwitterionic, protonated, deprotonated);
- Quinone form can drastically influence both the absorption and emission wavelengths and the fluorescence lifetime;
- During polymerization, different forms of dopamine units may exhibit different photochemical behavior, but their average effects mean that the individual fingerprints of each unit cannot be accurately identified

Acknowledgements:



Founding: **PN-III-P4-ID-PCE-2020-0770**



Research Team: **Dr. Alexandra FALAMAS; Dr. Anca PETRAN; Dr. Crina SOCACI;
Dr. Alex-Adrian FARCAŞ**



Data Center of INCDTIM; ORCA program developers



Dr. Jürgen Liebscher

Thank You for Your Attention